

Diamagnets and paramagnets



Curie law, Langevin and Brillouin function

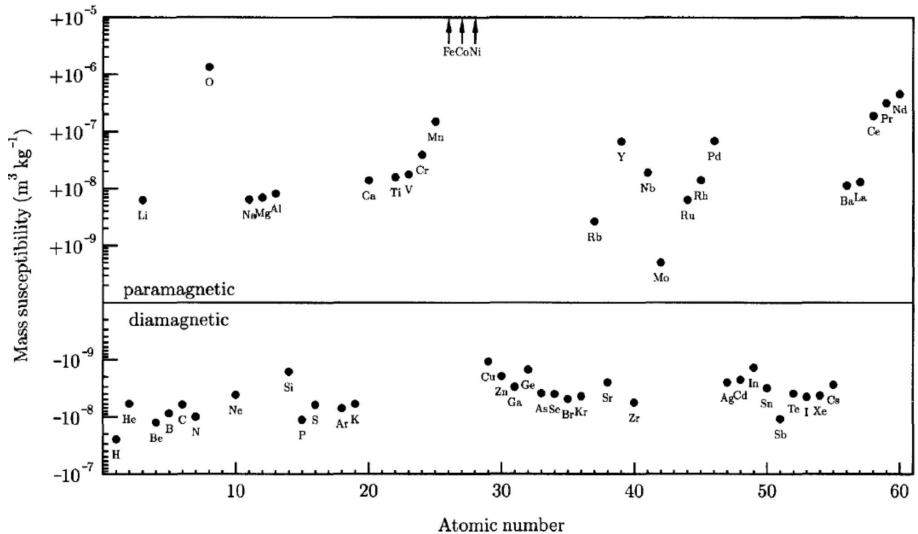


adiabatic demagnetization refrigeration



Pierre Curie and Paul Langevin





List of samples that showed diamagnetic response

Rock crystal.

Sulphate of lime.

Sulphate of baryta.

Sulphate of soda.

Sulphate of potassa.

Sulphate of magnesia.

Alum.

Nitric acid.

Sulphuric acid.

Muriatic acid.

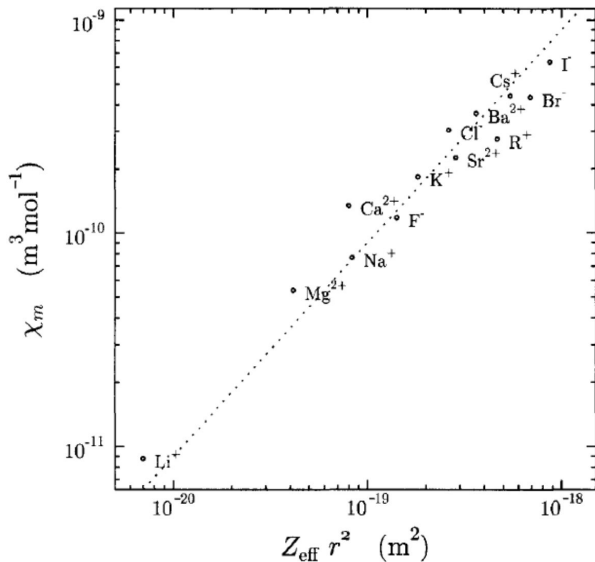
Solutions of various alkaline and
earthy salts.

Glass.

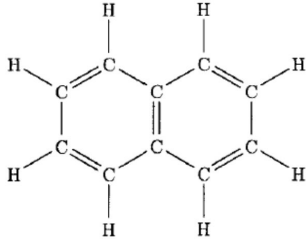
Litharge.

List of samples that showed diamagnetic response

Rock crystal.	Nitric acid.
Sulphate of lime.	Sulphuric acid.
Sulphate of baryta.	Muriatic acid.
Sulphate of soda.	Solutions of various alkaline and earthy salts.
Sulphate of potassa.	Glass.
Sulphate of magnesia.	Litharge.
Alum.	
Sugar.	Beef, dried.
Starch.	Blood, fresh.
Gum-arabic.	Blood, dried.
Wood.	Leather.
Ivory.	Apple.
Mutton, dried.	Bread.
Beef, fresh.	



$$\chi_{\text{dia}} \sim \langle r^2 \rangle$$

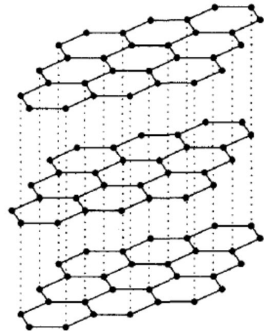


Naphtalene

$$\chi_{\perp} \simeq -40 \times 10^{-6} \text{ emu/g}$$

$$\chi_{\parallel} \simeq -187 \times 10^{-6} \text{ emu/g}$$

$$\chi_{\parallel}/\chi_{\perp} \simeq 4.5$$



Graphite

$$\chi_{\perp} \simeq -0.5 \times 10^{-6} \text{ emu/g}$$

$$\chi_{\parallel} \simeq -22 \times 10^{-6} \text{ emu/g}$$

$$\chi_{\parallel}/\chi_{\perp} \simeq 44$$

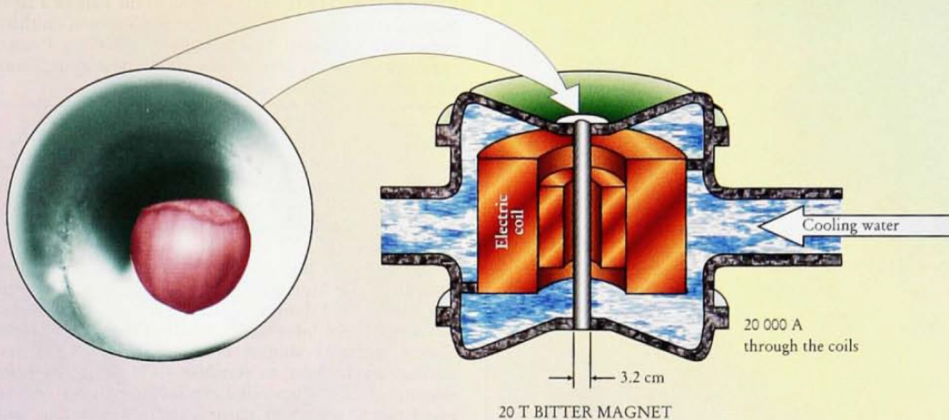
Atom	$\chi_{Di}/(1 \times 10^{-6} \text{ emu mol}^{-1})$	Atom	$\chi_{Di}/(1 \times 10^{-6} \text{ emu mol}^{-1})$
Ag	-31.0	C (ring)	-6.24
Al	-13.0	Ca	-15.9
As(III)	-20.9	Cl	-20.1
As(V)	-43.0	F	-6.3
B	-7.0	H	-2.93
Bi	-192.0	Hg(II)	-33.0
Br	-30.6	I	-44.6
C	-6.00	K	-18.5

χ_{Di}
(core)

λ_i

C=C	+5.5
C $\equiv$ C	+0.8
C=C-C=C	+10.6
Ar-C $\equiv$ C-Ar ^b	+3.85
CH ₂ =CH-CH ₂ -(allyl)	+4.5
C=O	+6.3

Cl-CR ₂ CR ₂ -Cl	+4.3
R ₂ CCl ₂	+1.44
RCHCl ₂	+6.43
C-Br	+4.1
Br-CR ₂ CR ₂ -Br	+6.24
C-I	+4.1



Levitating nuts

Applied field of 16 T generates the field of -1 G inside the nut

Diamagnetic levitation

Live frog



Water drop

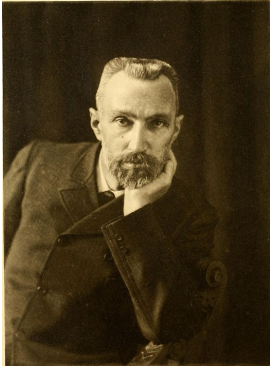


Effects of microgravity can be studied in the lab



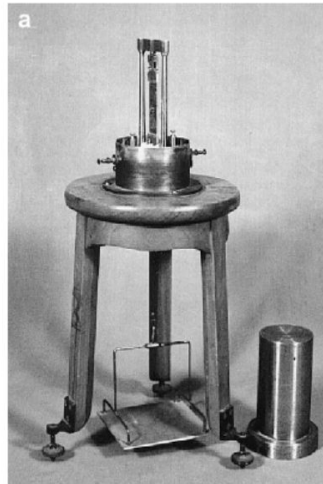
Person

Pierre Curie and Paul Langevin



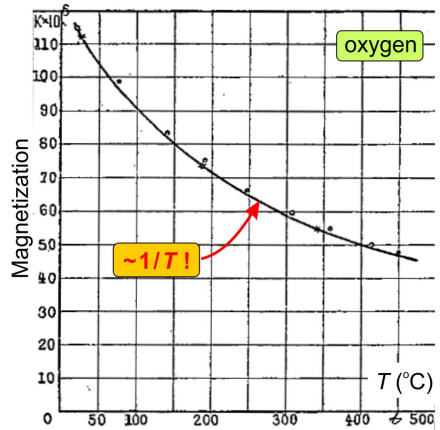
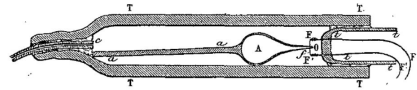
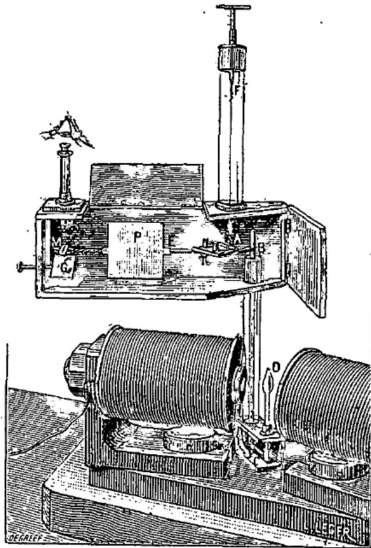
Pierre Curie
1859–1906

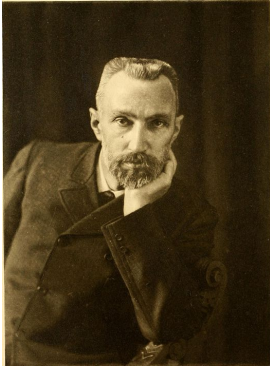
discovery of piezoelectricity
experimental studies
of T -dependent magnetic behavior



Strained quartz crystal used to measure currents on the pA scale

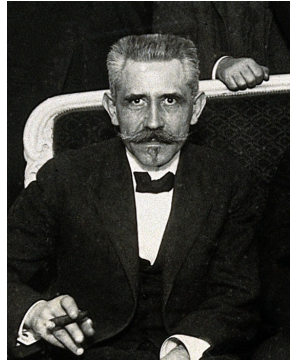
Curie's apparatus





Pierre Curie
1859–1906

discovery of piezoelectricity
experimental studies
of T -dependent magnetic behavior



Paul Langevin
1872–1946

use of piezoelectric sensors
for submarine detection
first theory of dia- and paramagnetism



Pierre Curie (1859–1906)

Marie Skłodowska-Curie (1867–1934)

1903: Nobel prize in physics for radioactivity

1911: Nobel prize in chemistry for the discovery of new chemical elements, especially radium (Marie only)

Curie's greatest discovery



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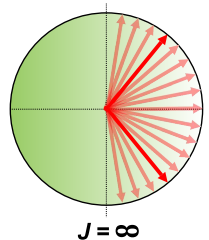
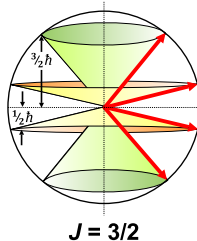
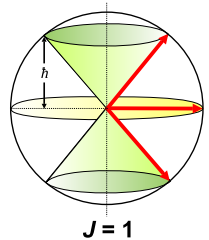
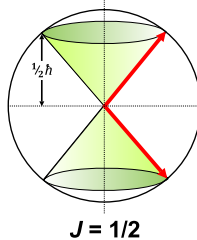
Langevin affair

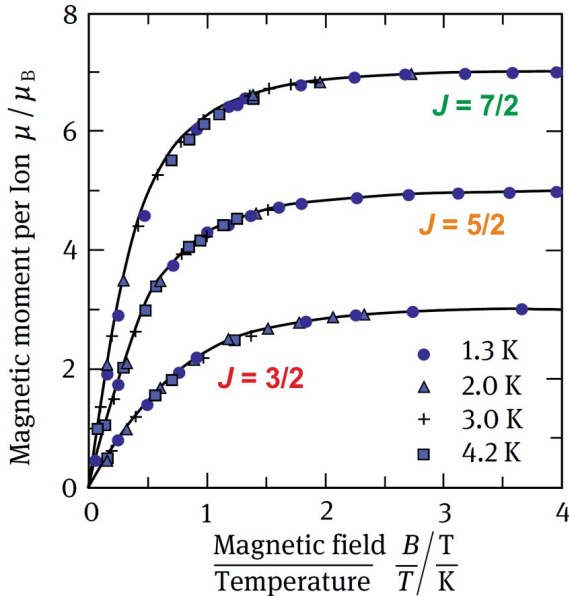




Léon Brillouin
1889–1968

PhD student of Langevin
better known for the Brillouin zone,
WKB approximation...





Universal behavior
as a function
of B/T



Material / Technology

adiabatic demagnetization refrigeration

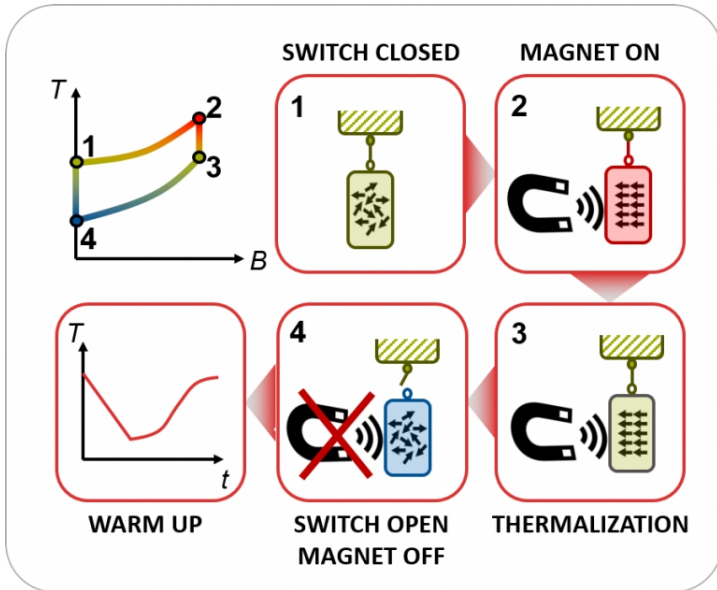


Image credit: cryodynamics.de

**16. *Einige Bemerkungen zur Magnetisierung
bei tiefer Temperatur;***

Ann. Phys. 386, 1154 (1926) *von P. Debye*

**A THERMODYNAMIC TREATMENT OF CERTAIN MAGNETIC
EFFECTS. A PROPOSED METHOD OF PRODUCING
TEMPERATURES CONSIDERABLY BELOW 1° ABSOLUTE**

BY W. F. GIAUQUE

JACS 49 1864 (1927)

RECEIVED DECEMBER 14, 1926

PUBLISHED AUGUST 5, 1927

Attainment of Temperatures Below 1° Absolute by Demagnetization of $\text{Gd}_2(\text{SO}_4)_3 \cdot 8\text{H}_2\text{O}$

Phys. Rev. 43, 768 (1933)

Department of Chemistry,
University of California,
Berkeley, California,
April 12, 1933.

W. F. GIAUQUE
D. P. MACDOUGALL

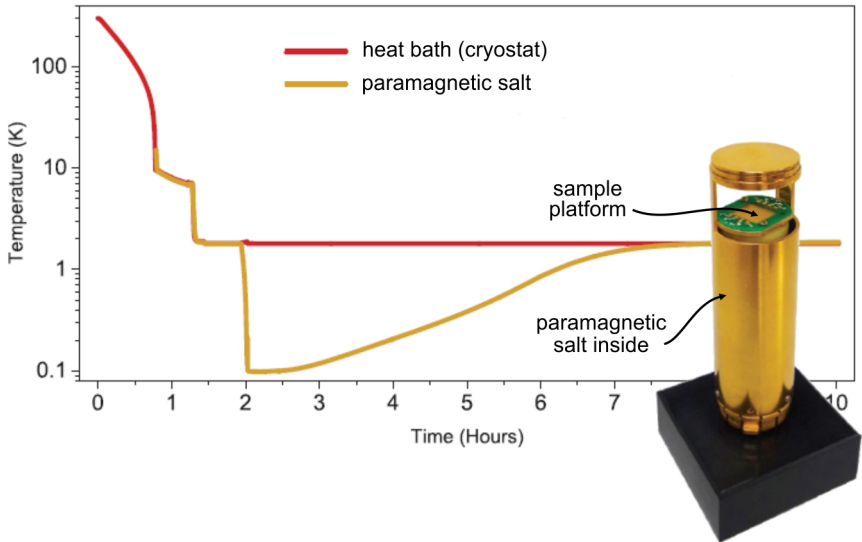
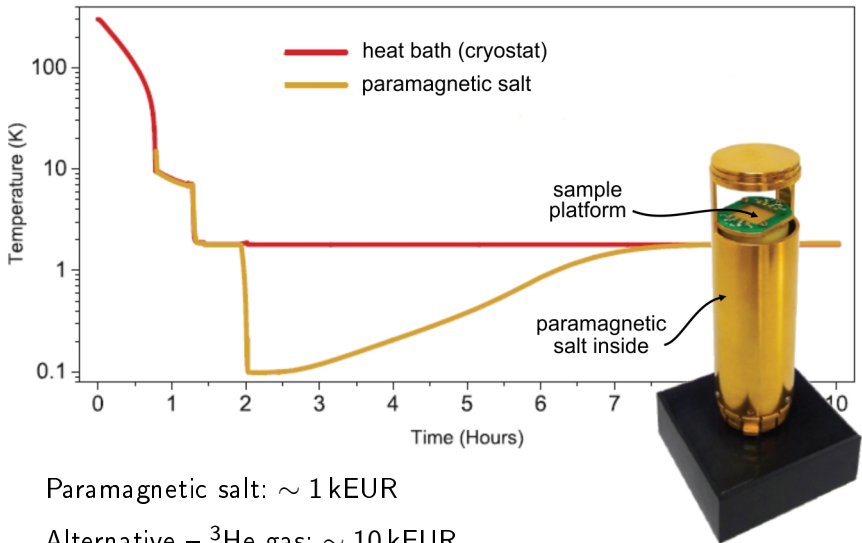


Image credit: Quantum Design (fair use)



Paramagnetic salt: ~ 1 kEUR

Alternative – ^3He gas: ~ 10 kEUR