

Diamagnets and paramagnets



Curie law, Langevin and Brillouin function

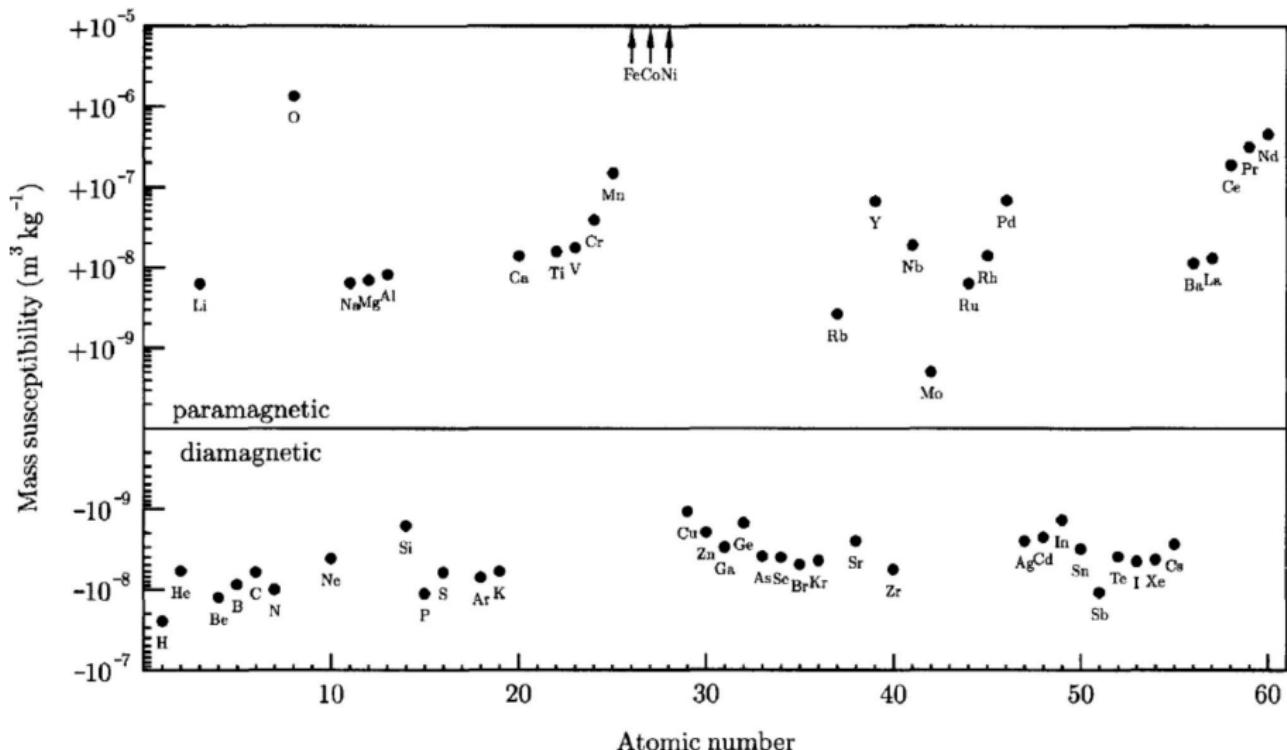


adiabatic demagnetization refrigeration



Pierre Curie and Paul Langevin





List of samples that showed diamagnetic response

Rock crystal.

Sulphate of lime.

Sulphate of baryta.

Sulphate of soda.

Sulphate of potassa.

Sulphate of magnesia.

Alum.

Nitric acid.

Sulphuric acid.

Muriatic acid.

Solutions of various alkaline and earthy salts.

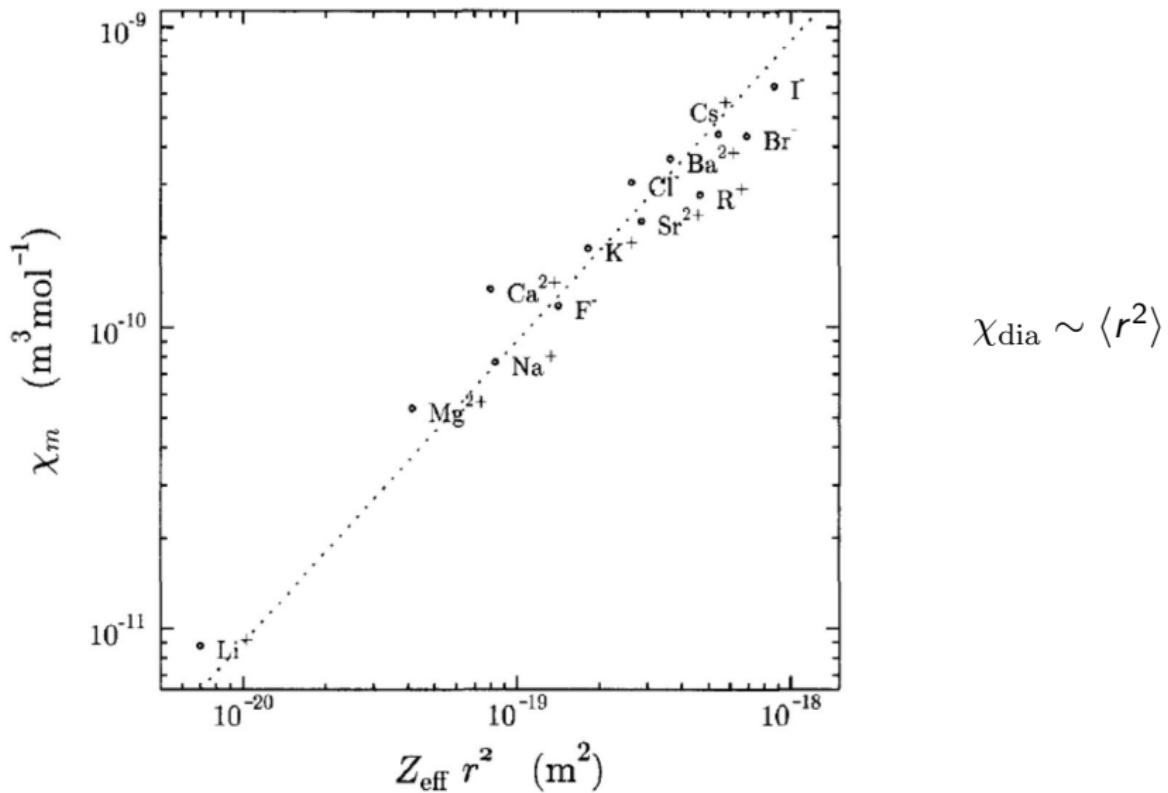
Glass.

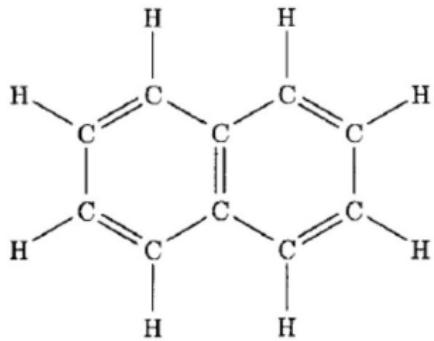
Litharge.

List of samples that showed diamagnetic response

Rock crystal.	Nitric acid.
Sulphate of lime.	Sulphuric acid.
Sulphate of baryta.	Muriatic acid.
Sulphate of soda.	Solutions of various alkaline and earthy salts.
Sulphate of potassa.	Glass.
Sulphate of magnesia.	Litharge.
Alum.	
Sugar.	Beef, dried.
Starch.	Blood, fresh.
Gum-arabic.	Blood, dried.
Wood.	Leather.
Ivory.	Apple.
Mutton, dried.	Bread.
Beef, fresh.	

Diamagnetic response



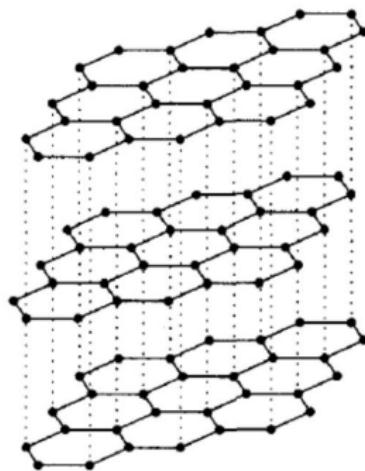


Naphthalene

$$\chi_{\perp} \simeq -40 \times 10^{-6} \text{ emu/g}$$

$$\chi_{\parallel} \simeq -187 \times 10^{-6} \text{ emu/g}$$

$$\chi_{\parallel}/\chi_{\perp} \simeq 4.5$$



Graphite

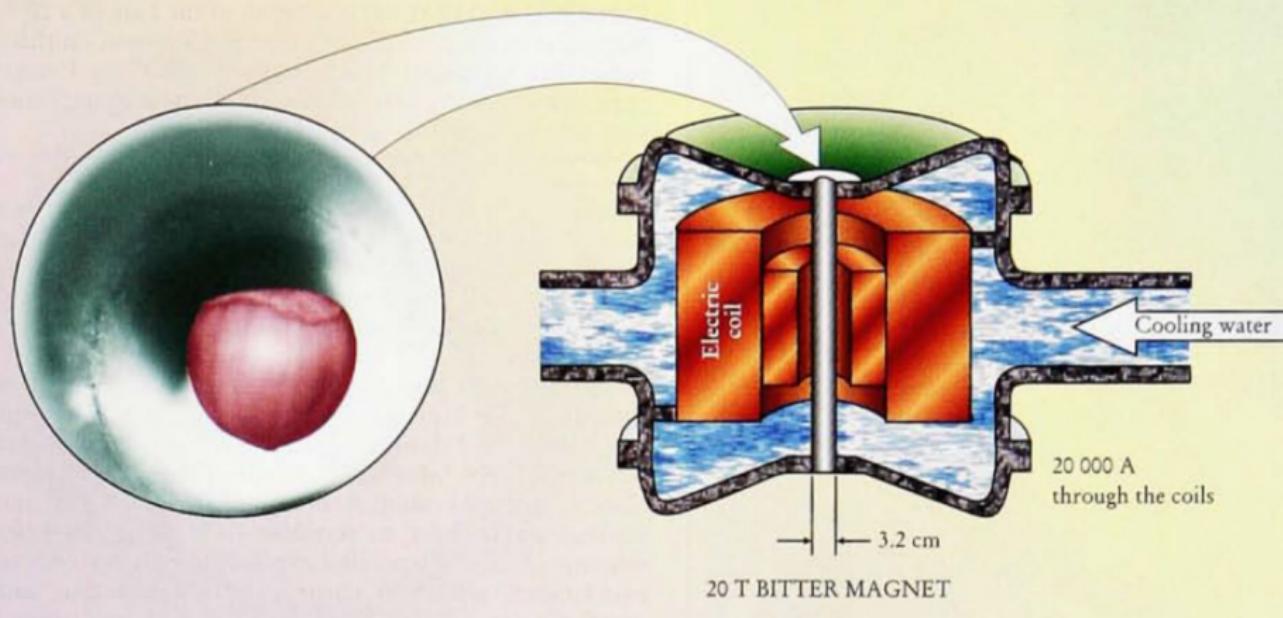
$$\chi_{\perp} \simeq -0.5 \times 10^{-6} \text{ emu/g}$$

$$\chi_{\parallel} \simeq -22 \times 10^{-6} \text{ emu/g}$$

$$\chi_{\parallel}/\chi_{\perp} \simeq 44$$

Atom	$\chi_{D_i}/(1 \times 10^{-6}$ emu mol ⁻¹)	Atom	$\chi_{D_i}/(1 \times 10^{-6}$ emu mol ⁻¹)	
Ag	-31.0	C (ring)	-6.24	
Al	-13.0	Ca	-15.9	
As(III)	-20.9	Cl	-20.1	
As(V)	-43.0	F	-6.3	
B	-7.0	H	-2.93	
Bi	-192.0	Hg(II)	-33.0	
Br	-30.6	I	-44.6	
C	-6.00	K	-18.5	
λ_i	C=C	+5.5	Cl-CR ₂ CR ₂ -Cl	+4.3
	C≡C	+0.8	R ₂ CCl ₂	+1.44
	C=C-C=C	+10.6	RCHCl ₂	+6.43
	Ar-C≡C-Ar ^b	+3.85	C-Br	+4.1
	CH ₂ =CH-CH ₂ -(allyl)	+4.5	Br-CR ₂ CR ₂ -Br	+6.24
	C=O	+6.3	C-I	+4.1

Diamagnetic levitation



Levitating nuts

Applied field of 16 T generates the field of -1G inside the nut

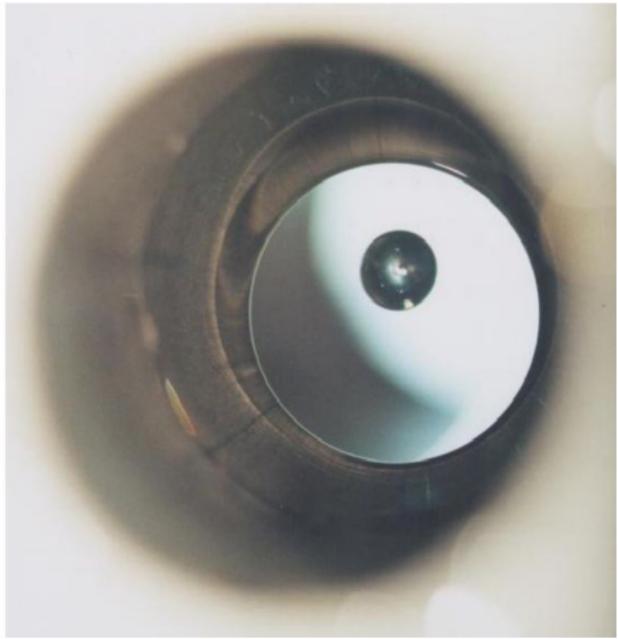
Physics Today 51(9), 36 (1998)

Diamagnetic levitation

Live frog



Water drop



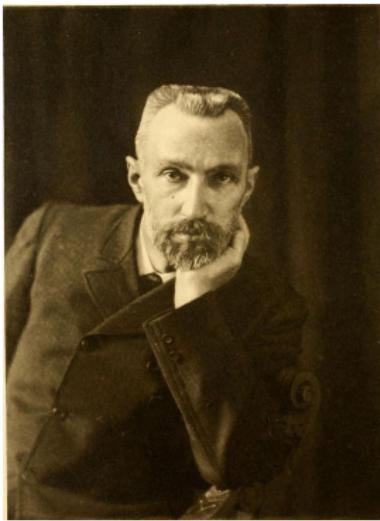
Effects of microgravity can be studied in the lab

Image credit: HfM L, Nijmegen



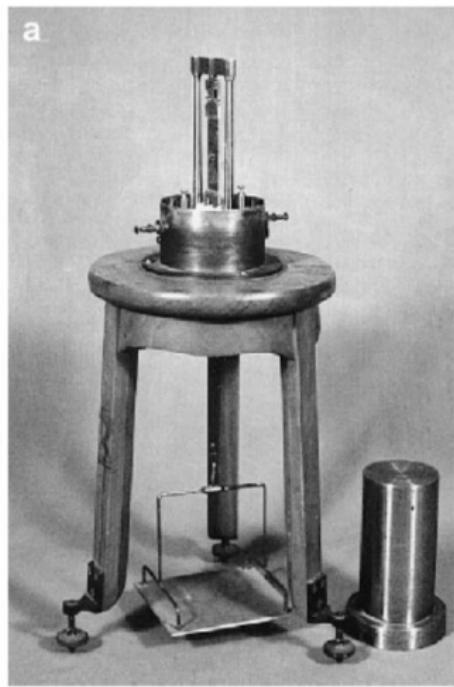
Person

Pierre Curie and Paul Langevin



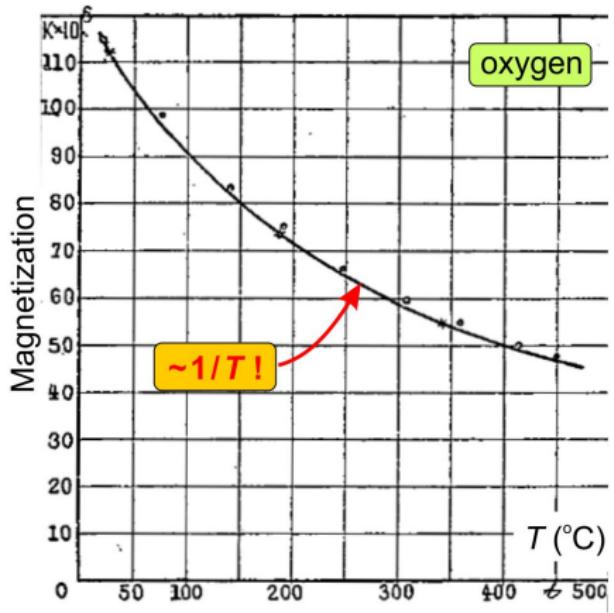
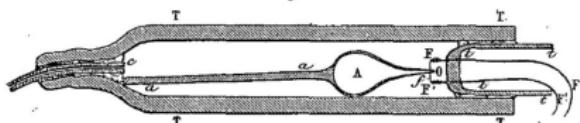
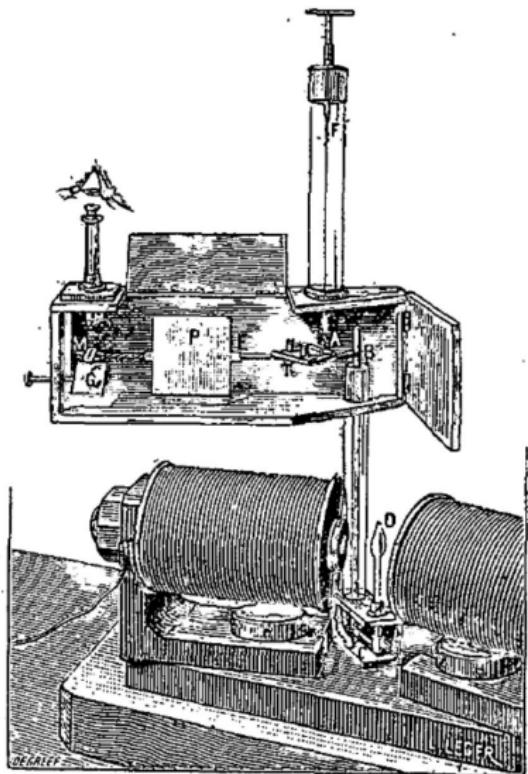
Pierre Curie
1859–1906

discovery of piezoelectricity
experimental studies
of T -dependent magnetic behavior

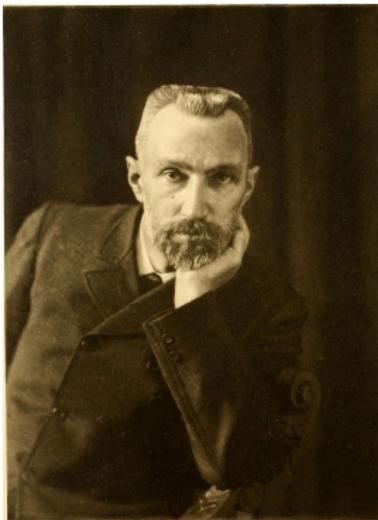


Strained quartz crystal used to measure currents on the pA scale

Curie's apparatus

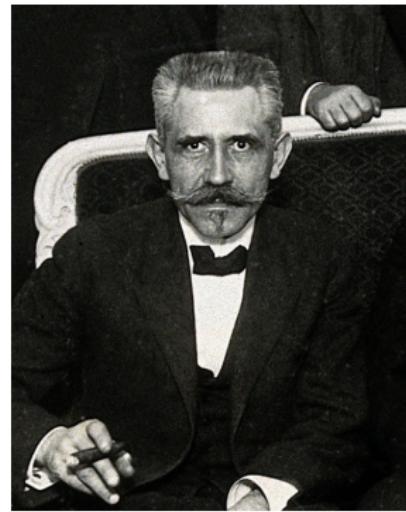


P. Curie, Ann. Chim. Phys. (5), 295 (1895)



Pierre Curie
1859–1906

discovery of piezoelectricity
experimental studies
of T -dependent magnetic behavior



Paul Langevin
1872–1946

use of piezoelectric sensors
for submarine detection
first theory of dia- and paramagnetism

Curie's greatest discovery



Pierre Curie (1859–1906)

Marie Skłodowska-Curie (1867–1934)

1903: Nobel prize in physics for radioactivity

1911: Nobel prize in chemistry for the discovery of new chemical elements, especially radium (Marie only)

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The Woman Who Found RADIUM but Lost LOVE

The Star Press (February 8, 1914)



Langevin affair

One does not need to be much of a scientist to understand where the trouble lies. In itself, the discovery of radium is one of the greatest achievements of modern science. But there were human failings involved. While the scientists working with radium were doing their best to keep it safe, they had to work in the basement of the university building. This was not the highest level of the academic world.

Her Researcher Error
As we have seen, Marie Curie was the first person to discover radium. She did this by using a process called radiotherapy. This involved exposing the body to high levels of radiation. This was done to treat cancer.



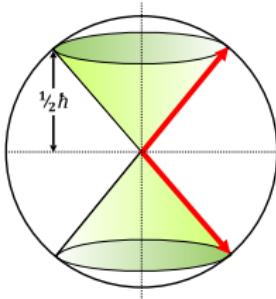


Léon Brillouin

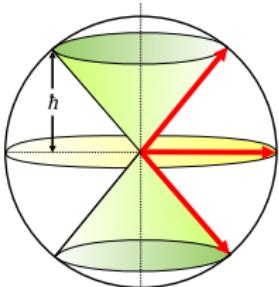
1889–1968

PhD student of Langevin

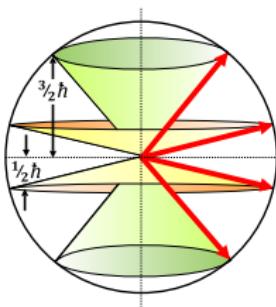
better known for the Brillouin zone,
WKB approximation...



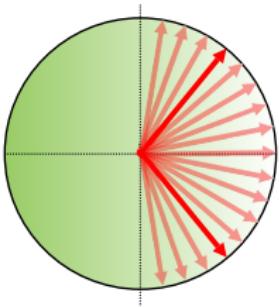
$$J = 1/2$$



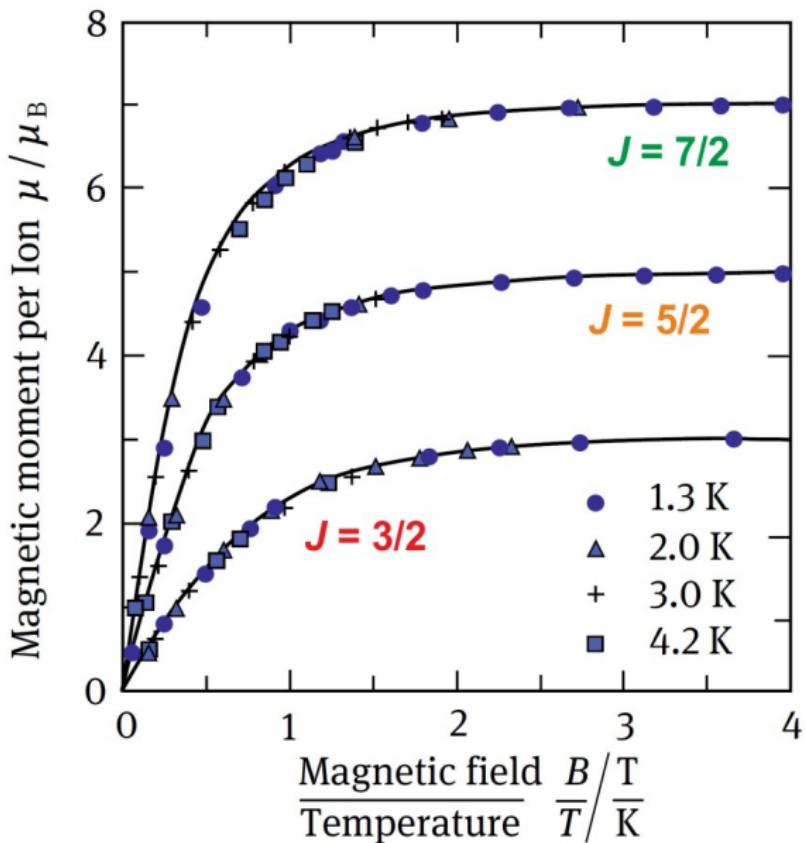
$$J = 1$$



$$J = 3/2$$



$$J = \infty$$



Universal behavior
as a function
of B/T



Material / Technology

adiabatic demagnetization refrigeration

Operation principle

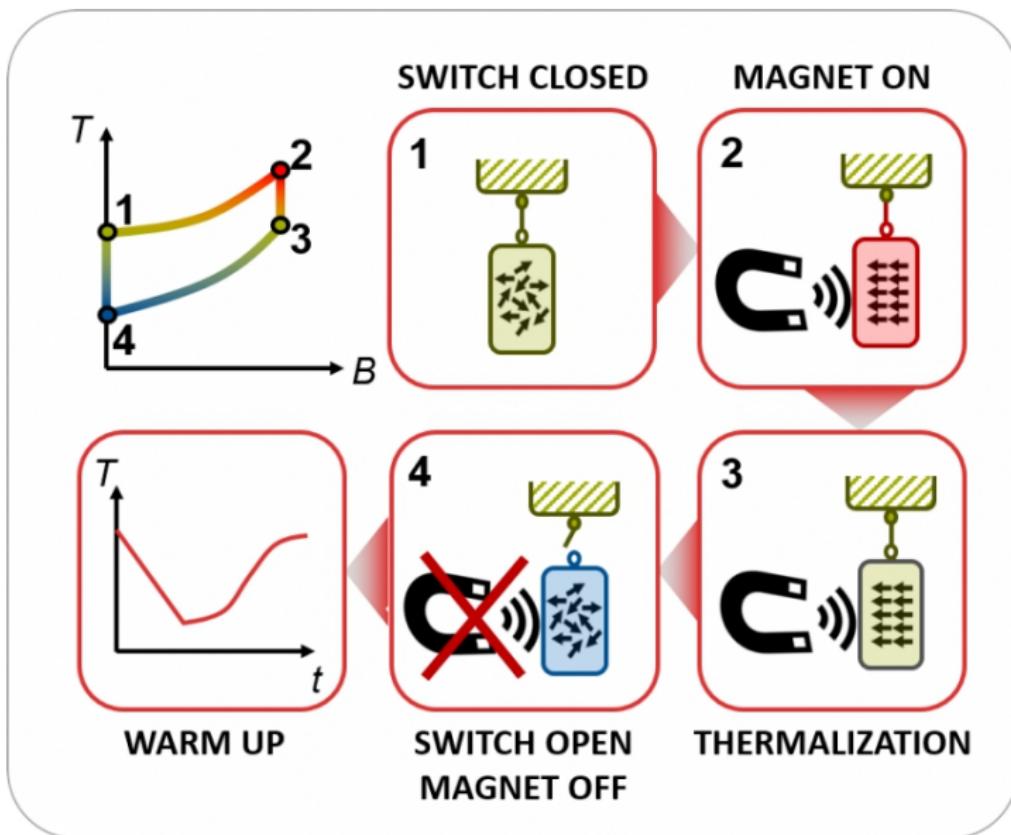


Image credit: cryodynamics.de

16. *Einige Bemerkungen zur Magnetisierung bei tiefer Temperatur;*
Ann. Phys. 386, 1154 (1926) von P. Debye

A THERMODYNAMIC TREATMENT OF CERTAIN MAGNETIC EFFECTS. A PROPOSED METHOD OF PRODUCING TEMPERATURES CONSIDERABLY BELOW 1° ABSOLUTE

BY W. F. GIAUQUE

JACS 49 1864 (1927)

RECEIVED DECEMBER 14, 1926

PUBLISHED AUGUST 5, 1927

Attainment of Temperatures Below 1° Absolute by Demagnetization of $\text{Gd}_2(\text{SO}_4)_3 \cdot 8\text{H}_2\text{O}$

Phys. Rev. 43, 768 (1933)

Department of Chemistry,
University of California,
Berkeley, California,
April 12, 1933.

W. F. GIAUQUE
D. P. MACDOUGALL

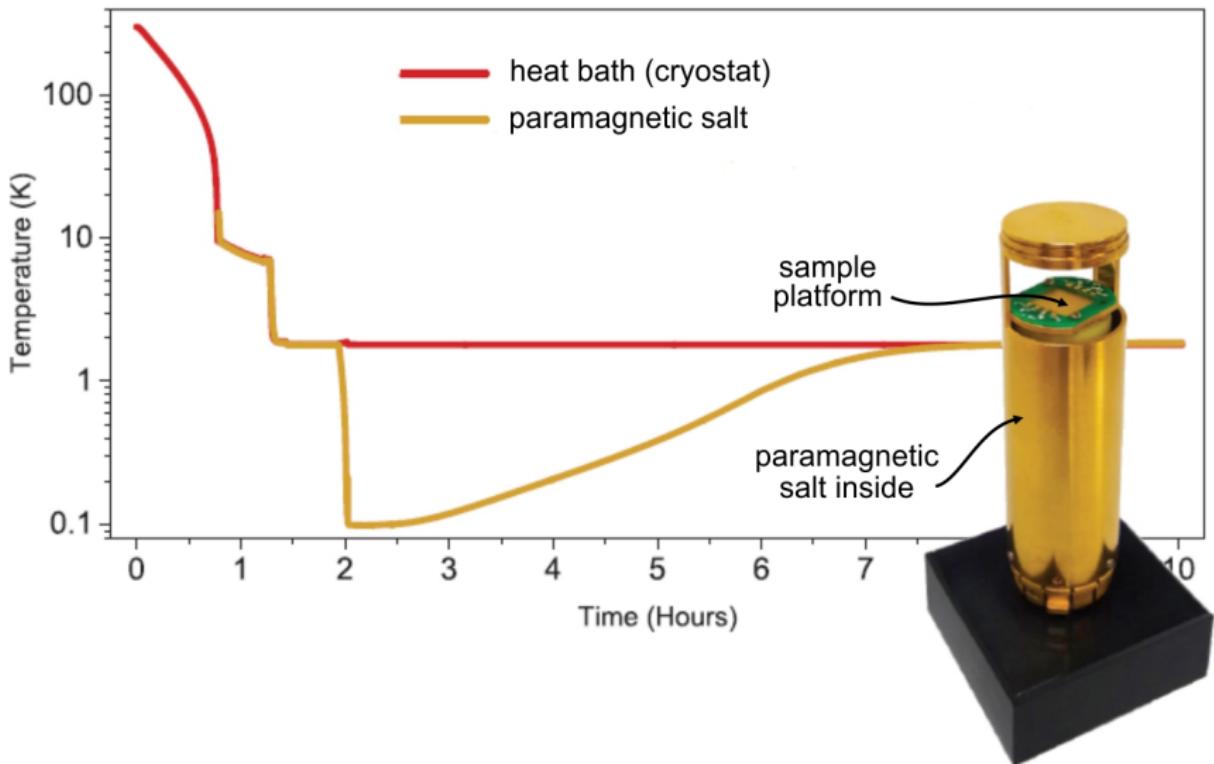


Image credit: Quantum Design (fair use)

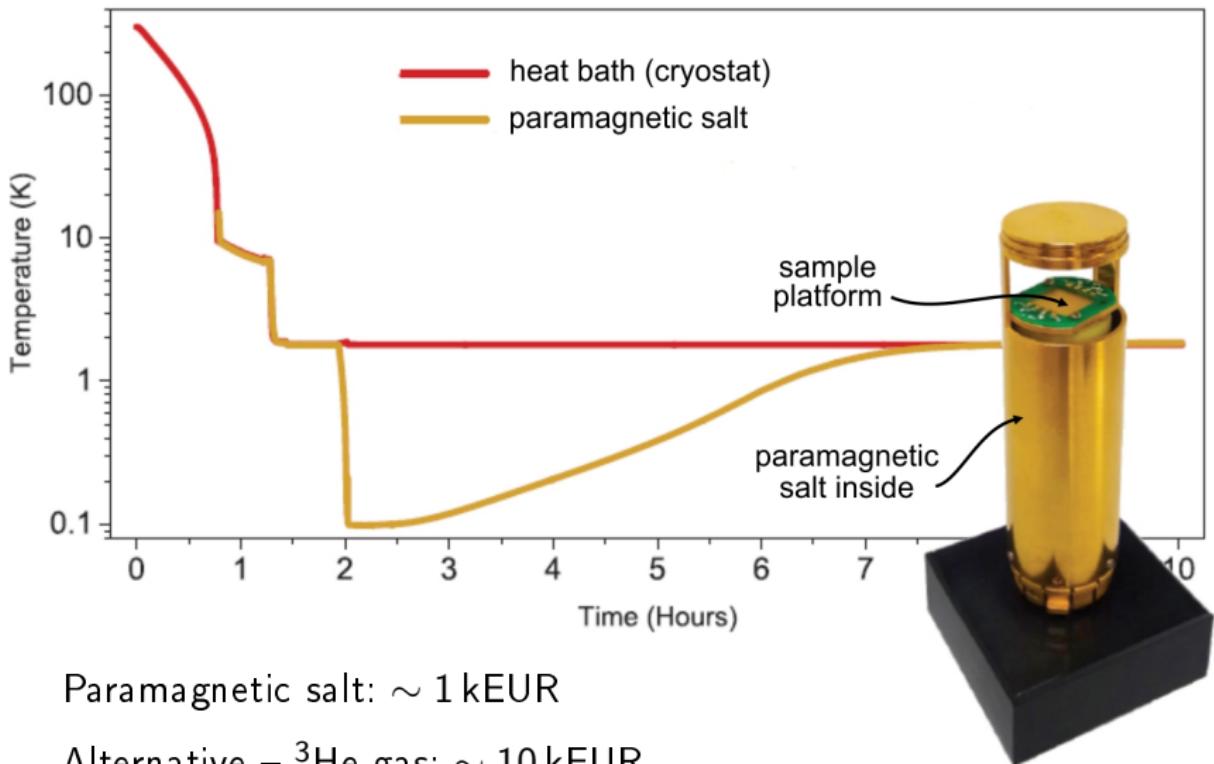


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