

Magnetism of metals



Pauli paramagnetism, Landau diamagnetism
Stoner instability



de Haas - van Alphen effect



Wolfgang Pauli





Person

Wolfgang Pauli

Pauli principle and Pauli effect

Pauli principle: two electrons can not occupy the same state



Wolfgang Pauli

1900–1958

1945 Nobel prize in physics for
"his discovery of a new law of Nature,
the exclusion principle or Pauli principle"

Pauli principle: two electrons can not occupy the same state

Pauli effect: a functioning device and Wolfgang Pauli may not occupy the same room
(second Pauli exclusion principle)



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“Das ist nicht nur nicht richtig,
es ist nicht einmal falsch!”

W. Pauli



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the exclusion principle or Pauli principle"

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"When I die, my first question to the devil will be:
What is the meaning of the fine structure constant?"

W. Pauli
 $\alpha \simeq 1/137$



Wolfgang Pauli
1900–1958
1945 Nobel prize in physics for "his discovery of a new law of Nature, the exclusion principle or Pauli principle"

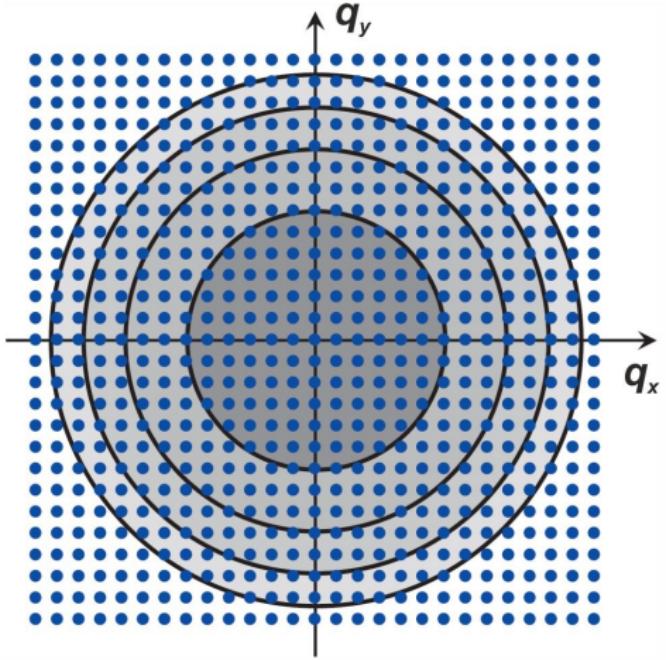
Über Gasentartung und Paramagnetismus.

Von W. Pauli jr. in Hamburg.

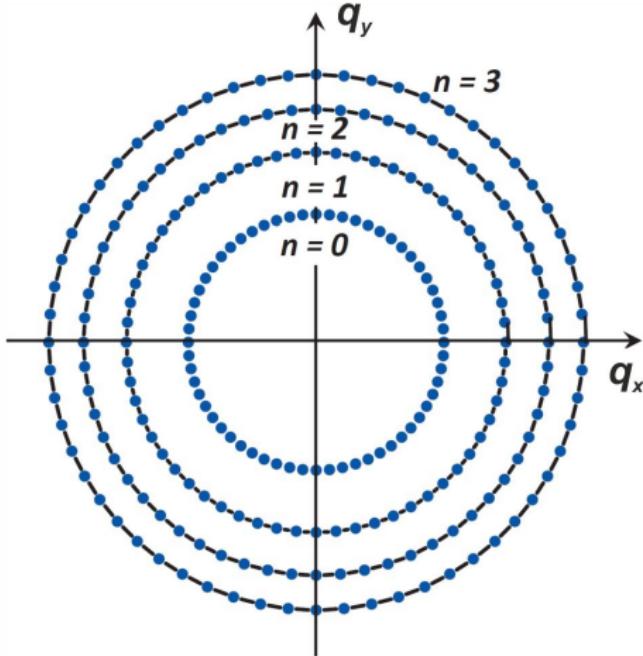
(Eingegangen am 16. Dezember 1926.)

	Na	K	Rb	Cs
calc. $(\chi_0)_{\text{ber.}}$	$= (6,57;$	$5,2;$	$4,88;$	$4,54) \cdot 10^{-7}$
exp. $(\chi_0)_{\text{beob.}}$	$= (5,8$	$; 5,1;$	$0,6$	$; -0,5$

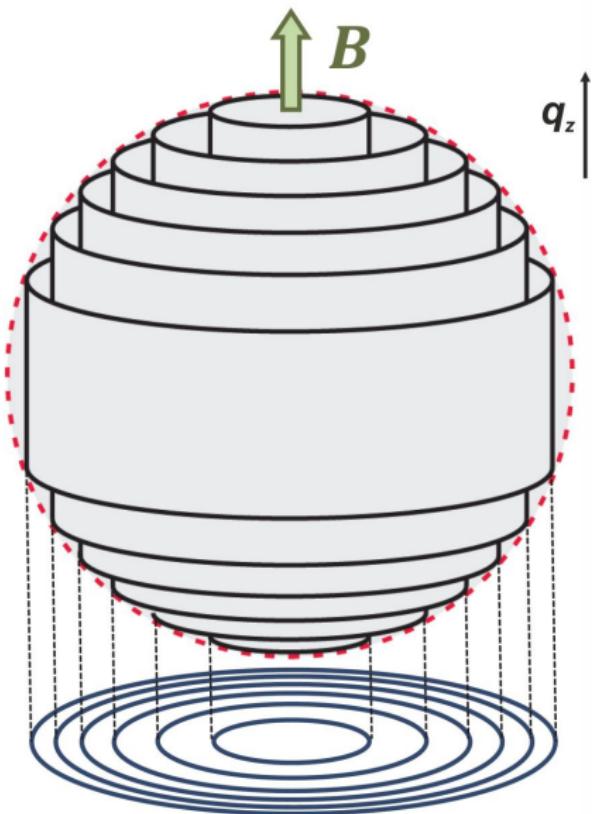
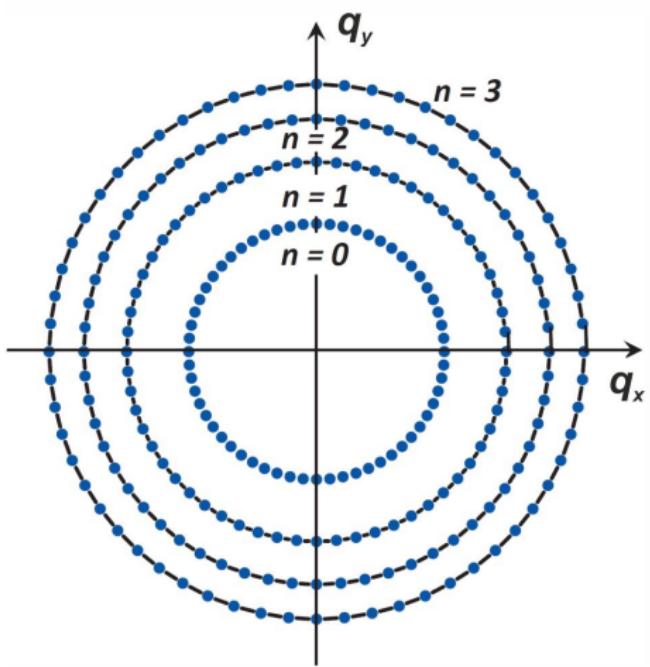
zero field ($B = 0$)



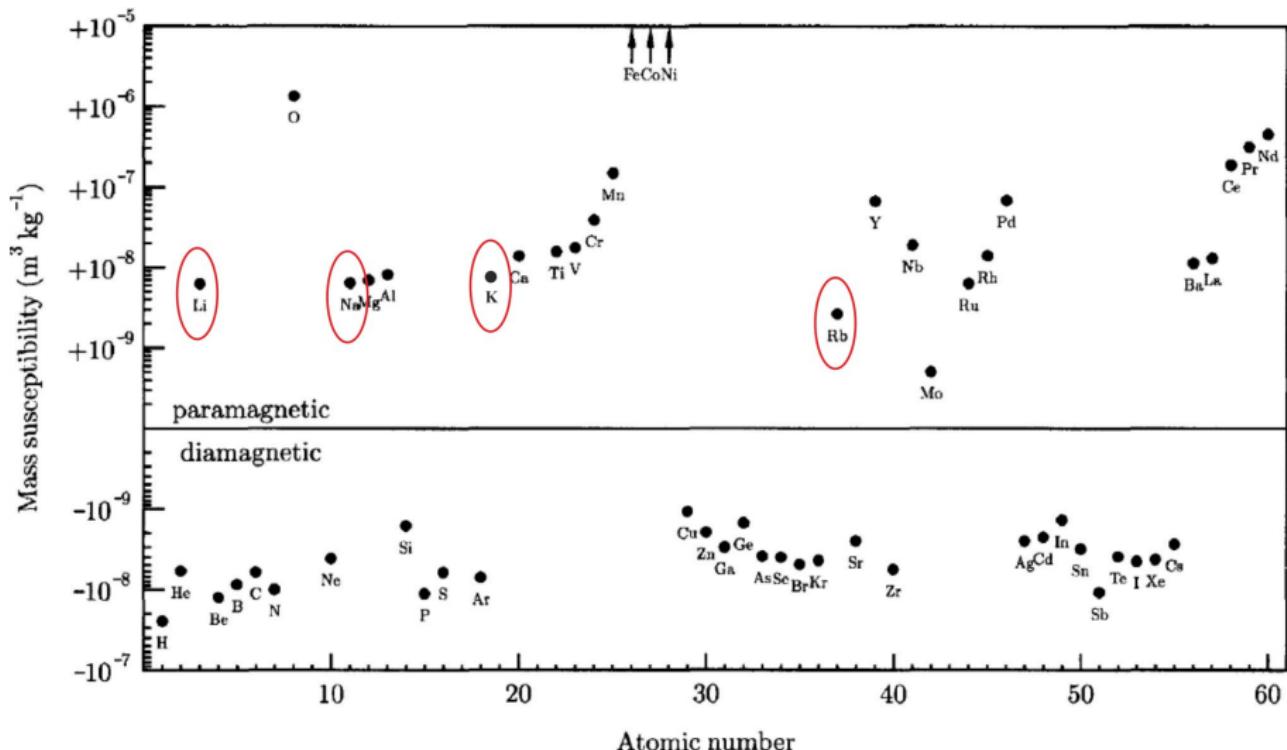
magnetic field ($B \neq 0$)



Magnetic field re-distributes electronic states of a crystal



Paramagnetic vs. diamagnetic metals

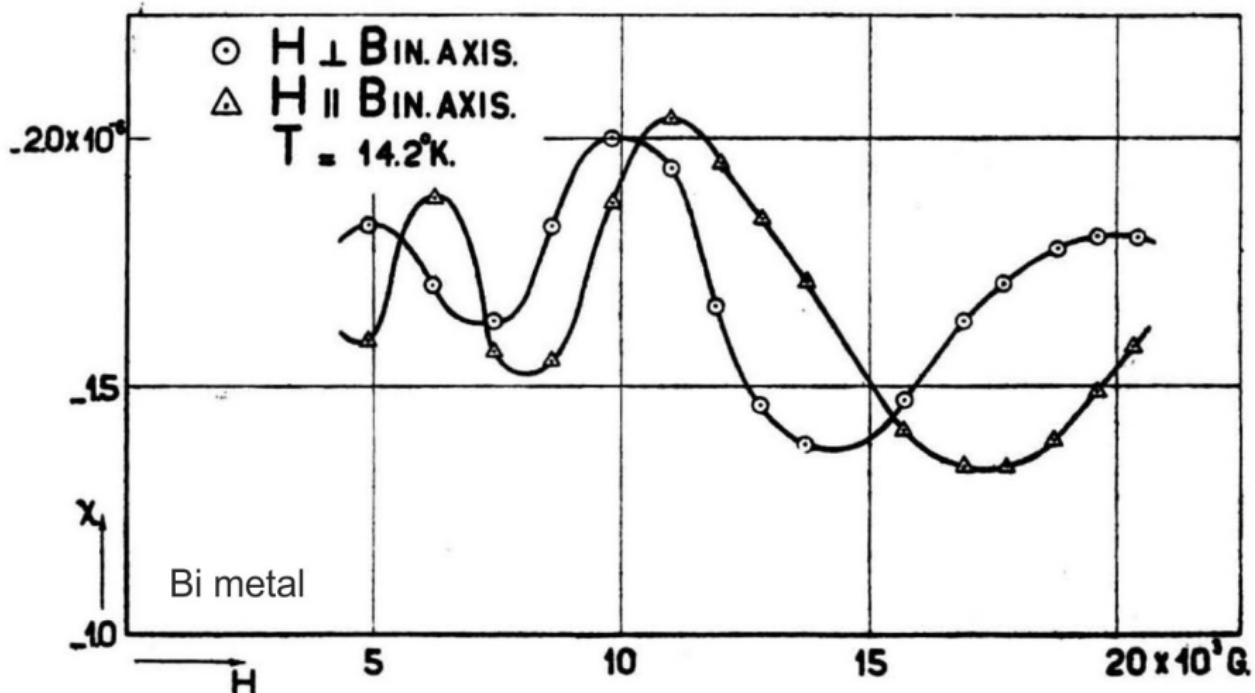


S. Blundell, Magnetism in Condensed Matter
with corrections based on CRC Handbook of Chemistry and Physics

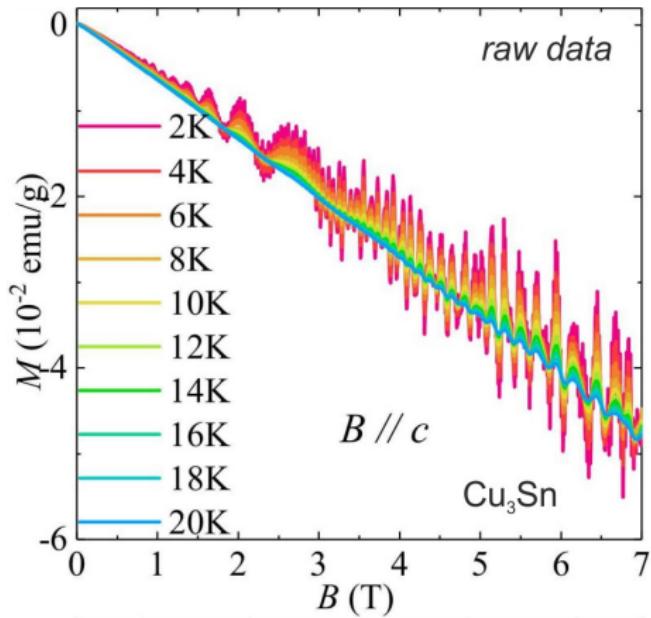


Experiment

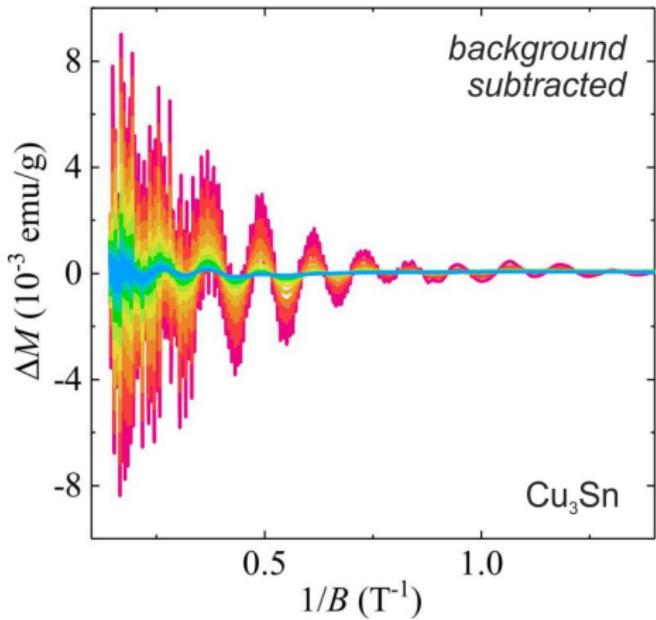
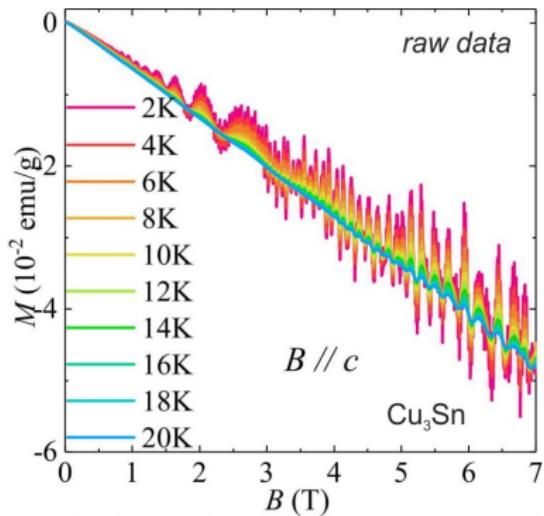
de Haas - van Alphen effect



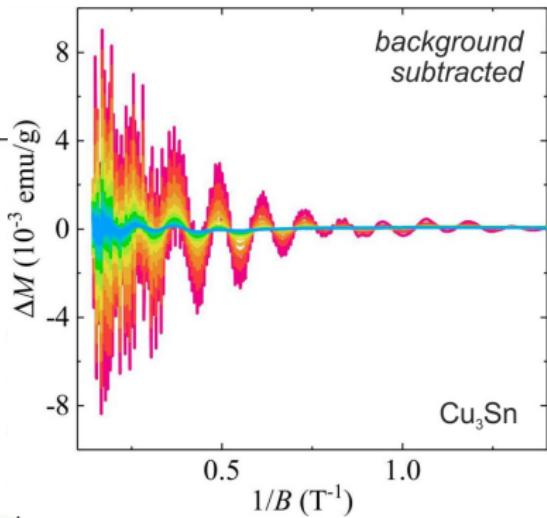
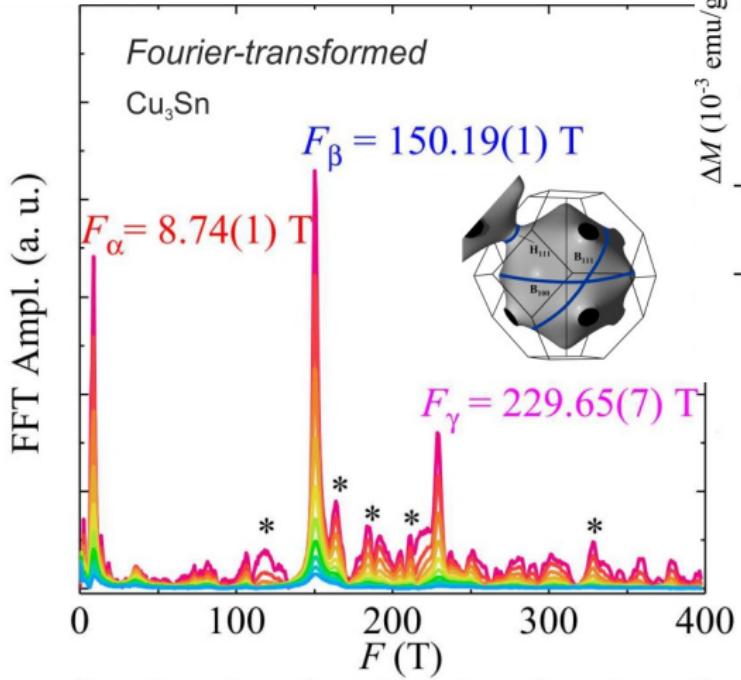
Getting the oscillations



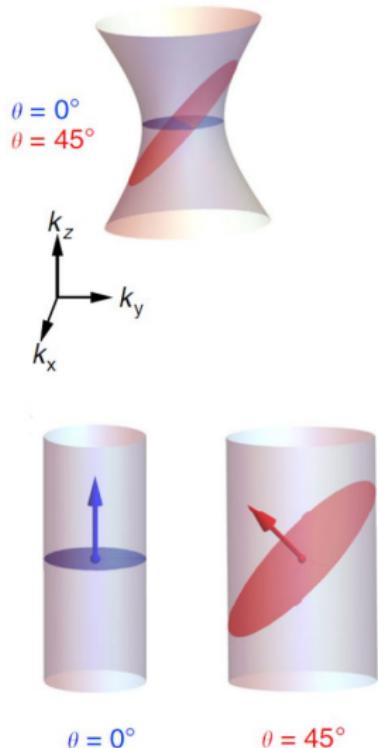
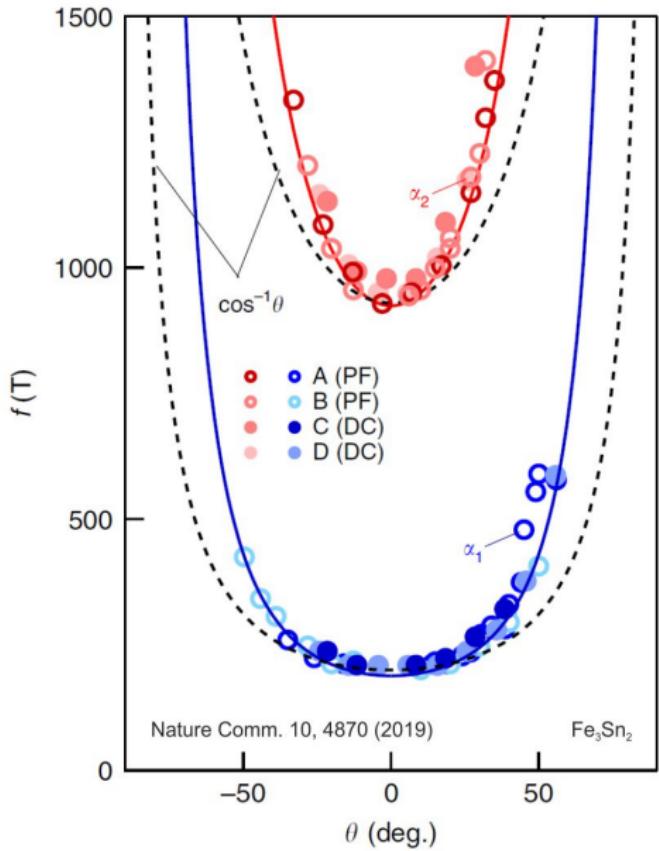
Getting the oscillations



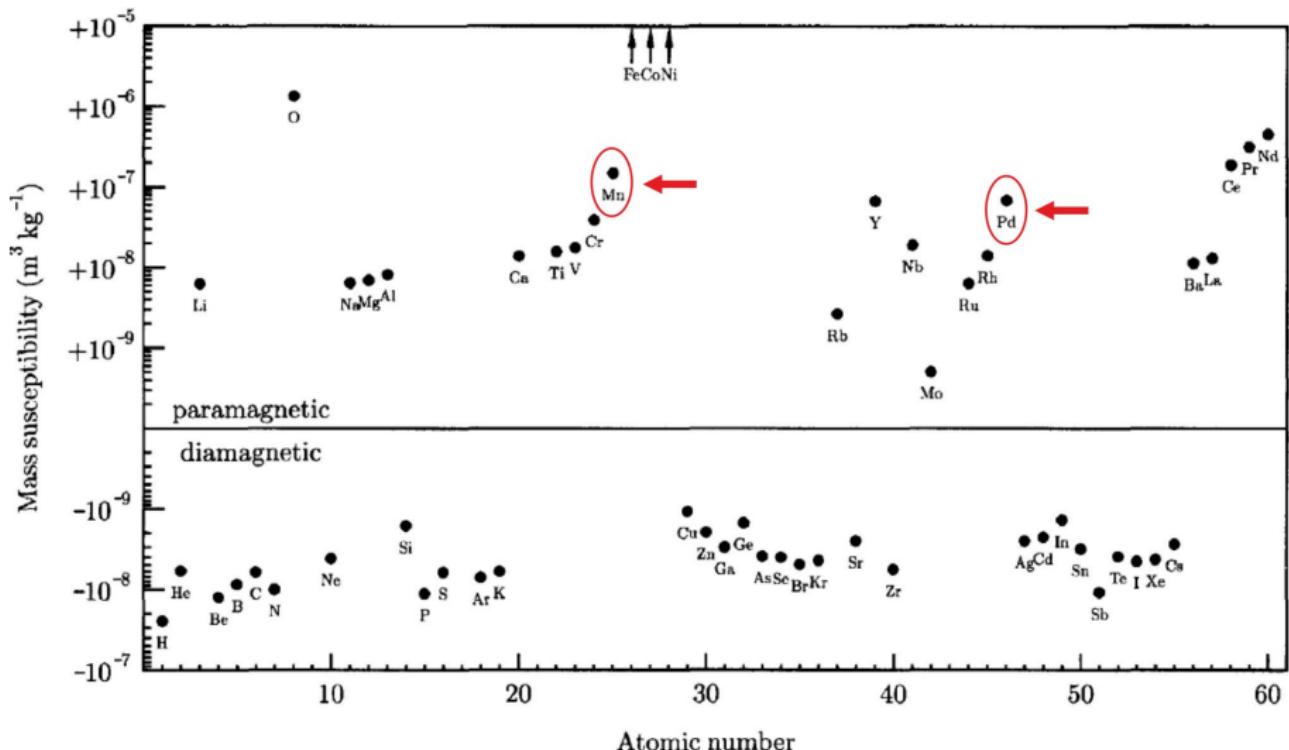
Getting the oscillations



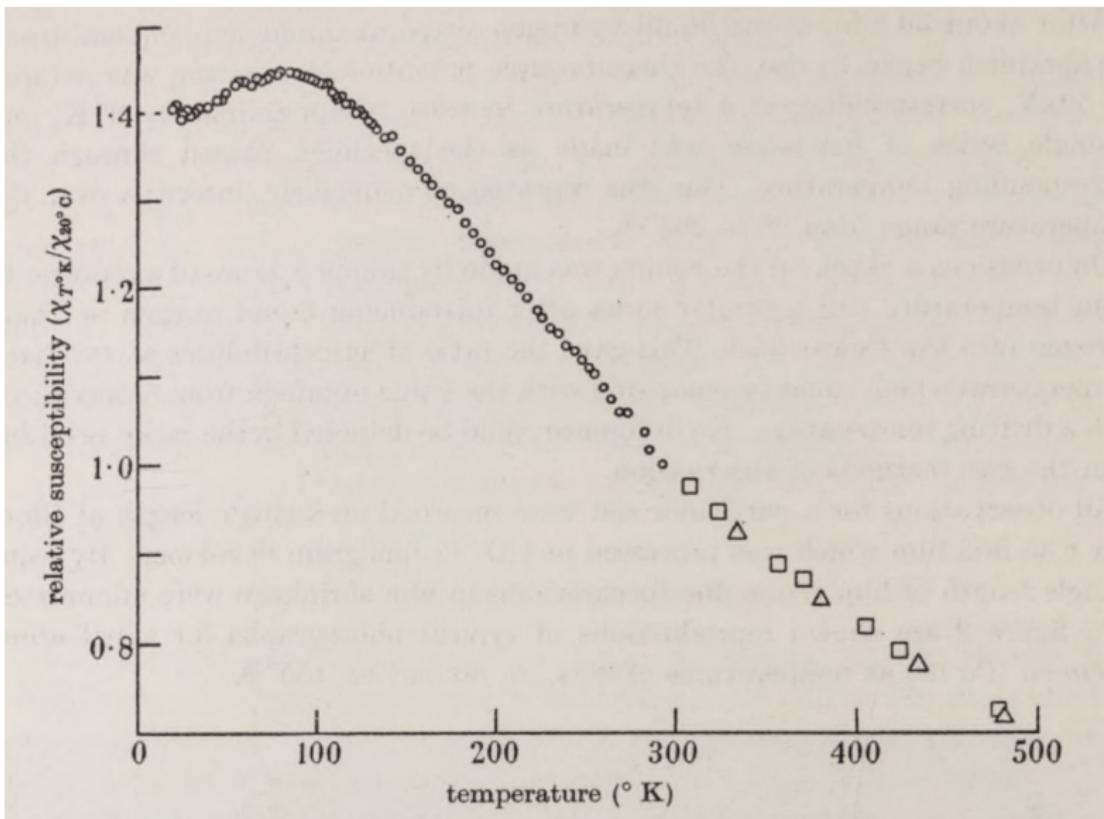
Angular dependence: full Fermi surface



Stoner enhancement



Magnetism of palladium



Proc. Royal Soc. A 212, 137 (1952)